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1) Fill in the chart above.

2) What is Law of conservation of mass and how does it relate to chemical

reactions? 3) What are physical properties? Give 3

examples. 4) What are chemical

properties? Give 3 examples. 5) Name

and locate the five

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periodic families 6)
Which group (family)
is the least reactive?
Most reactive?

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-The number of significant digits to which a value has been reliably measured. -The more decimal places, the more precise the measurement. -The greater the ...

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Summary of first
semester reactions.
Summary of reactions

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Reactions of different functional groups (looks like Orgo threw up)

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Substances that make up all matter.

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Energy Levels and
Orbitals A lot of
chemistry is explained
by the sharing and
trading of electrons
between atoms.
Understanding how
electrons are

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arranged in an atom is a building block of Chem I. Electrons in an atom are contained in specific energy levels (1, 2, 3, and so on) that are different distances from the nucleus.

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Questions (8%) Given
the element, atomic
mass, and ionic

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charge, be able to find the protons, neutrons, and electrons.

Understand the valence electrons determine a substance's chemical properties. Know the definition of an isotope. Given the element, be able to identify the electron configuration.

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your textbook's
chapter review or on
the study guide.

10 Study Pointers for
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Chemists need to
write out formulas and
equations to solve
chemistry problems.
In this unit, we begin
to name and write

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Compounds and learn how to write out and balance chemical equations. Equations enable us to describe chemistry topics in mathematical terms and predict the outcomes of reactions.

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