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A somewhat outdated way to refer to a solute is as the tonic (a phrase we hold onto in phrases like “gin and tonic,” where the tonic is dissolved in the gin). Using that, we wind up with two more descriptions of the situation above: Side 1 is hypertonic to

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side 2. That means that the solution on side 1 has more solute dissolved in it than the solution on side 2.

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(Section 5-1) Answer
Sheet

OSMOSIS is the
diffusion of water
through a cell
membrane. When
water goes through

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the cell membrane, no matter what direction, in or out, it is called "osmosis". In class we observed osmosis in plant cell by: FIRST putting a drop of water on a slide. NYS REGENT DIFFUSION LAB - The Living Osmosis. Regents Biology Diffusion And Osmosis Answer Key

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a special type of
diffusion. It is the

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diffusion of solvent or
water across a semi-
permeable membrane
(a membrane that
allows for the
diffusion of certain
solutes and water)
from an area of
higher

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Components

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Osmosis is the diffusion of water across a membrane in response to osmotic pressure caused by an imbalance of molecules on either side of the membrane.

Osmoregulation is the process of maintenance of salt and water balance (osmotic balance)

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across membranes
within the body ' s
fluids, which are
composed of water
plus electrolytes and
non ...

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O – 90% Beaker H 2
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movement below. Cell
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beaker is hypertonic
and the cell is
hypotonic. The water
will flow out of the
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You ...

Adjust the concentration of a solute on either side of a membrane in a cell and observe the system as it adjusts to the conditions through osmosis. The initial concentration of the solute can be

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manipulated, along with the volume of the cell.

Osmosis Gizmo :
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Osmosis. Imagine you have a cup that has (100 mL) water, and you add (15 g) of table sugar to the water. The sugar dissolves and the

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mixture that is now in the cup is made up of a solute (the sugar) that is dissolved in the solvent (the water). The mixture of a solute in a solvent is called a solution.. Imagine now that you have a second cup with (100 mL) of ...

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